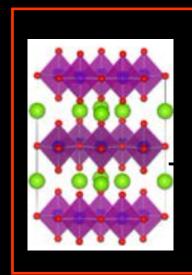
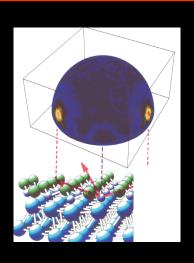
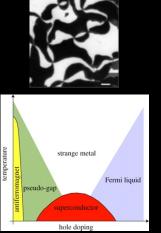
Homochirality and information content of the building blocks of life via theoretical spectroscopy

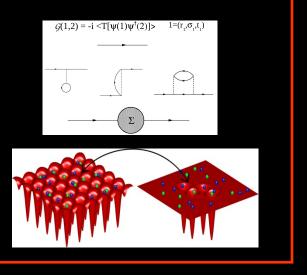
Fabiana Da Pieve

Laboratoire des Solides Irradiés, École Polytechnique, Palaiseau, France











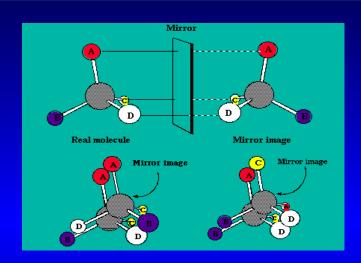


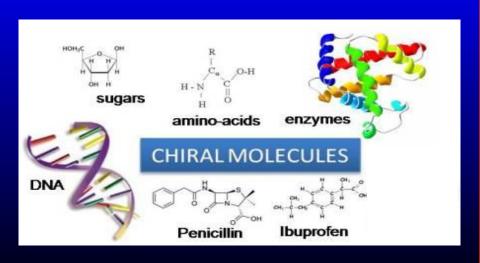


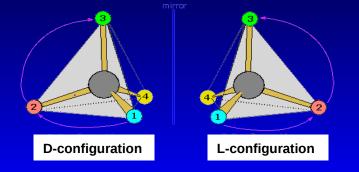




The homochirality of the building blocks of life (DNA, sugars, amino acids) is unexplained







Biological homochirality: L-amino acids and D-sugars

Origin?

P-breaking mechanism

exclude EM interaction

weak interaction? No, too weak

a chemical reasoning

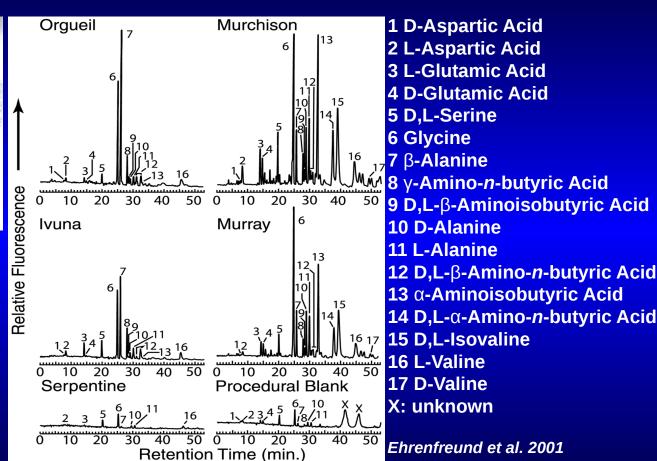
bifurcation when big chain polymers form?

Amino acids found in meteorites, with a preferred chirality



Excess in one mirror image

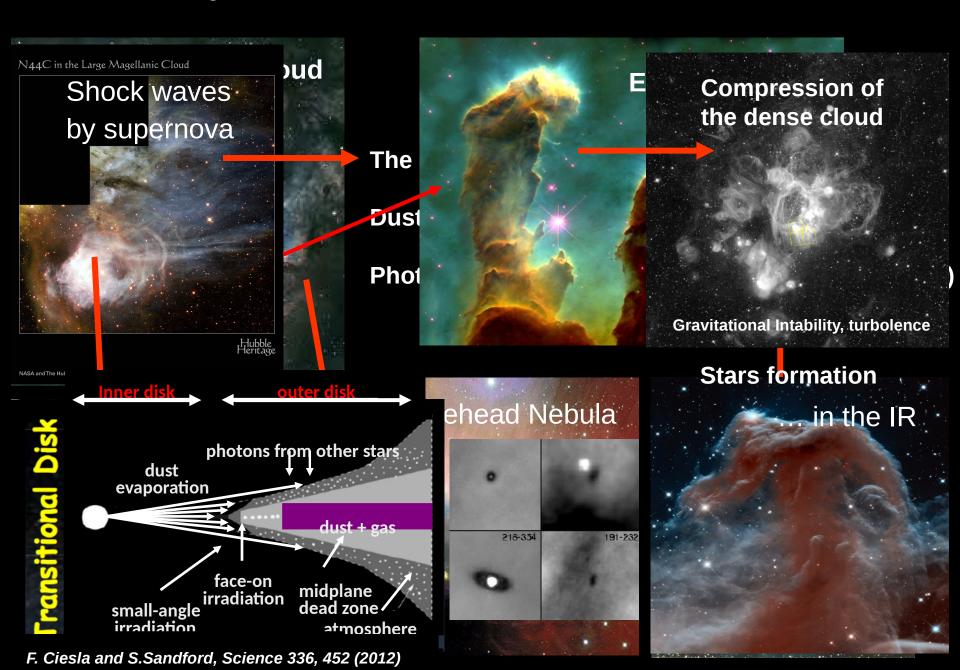
7-9% up to 60%



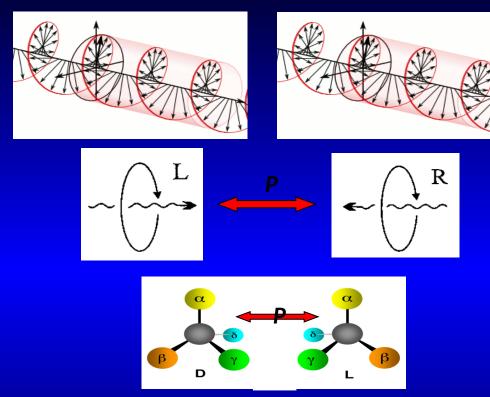
Influence of a chiral cosmic perturbation?

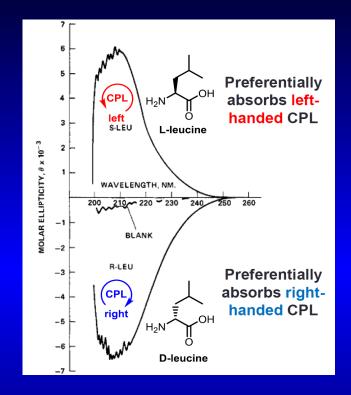
How amino acids formed in a cosmic context?

History of a meteorite linked to interstellar Ices



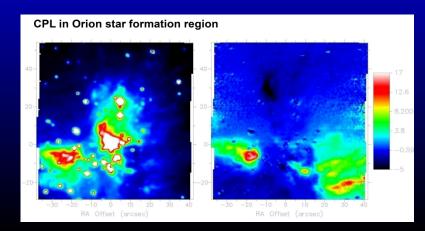
P-breaking by interaction with chiral perturbations: circularly polarized light (CPL) by close-by young stars





Cat's nebula 32% (IR), quadrupolar pattern,
Orion star formation region
O stars: X-ray polarimetry
black hole@center of our galaxy sgrA* (Radio)

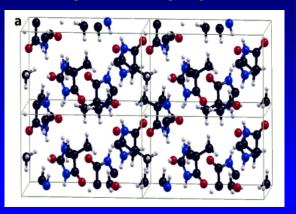
Kwon et al., Astrophysical Journal Letter 765, L 6 (2013) Bailey et al, Science 281, 672 (1998) Sudnik et al, AIP Conf. Proc. 1429,114 (2012) Munoz et al., The Astrophysical Journal 745, 115 (2012)



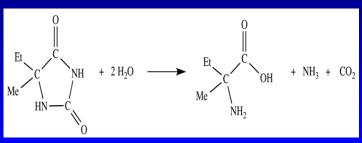
Imagine a "cold - warming phase - back to cold" scenario:

from precursors in molecular clouds to amino acids in outer-disk small bodies

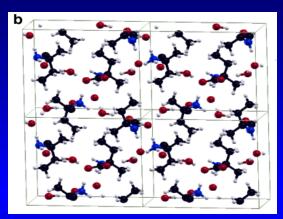
1) precursor, cold environment 5-ethyl-5-methylhydantoin



2) warming phase, formation of the amino acid



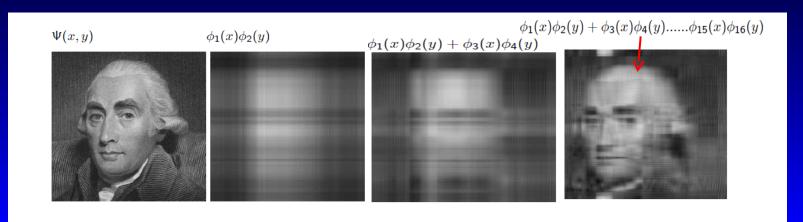
3) Isovaline, recondensation within ice



Search for strong peaks in the difference of absorption for left and right circular polarization, in correspondance of energy needed to break specific C-C bonds

Which theoretical approach?

Wave-function approaches: unpractical in solids

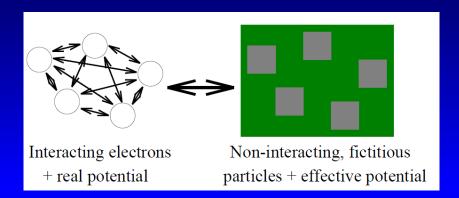


Propagators, Density -based approaches

$$\Psi(\mathbf{r}_1,\mathbf{r}_2,..,\mathbf{r}_N,t) \longrightarrow G(\mathbf{r}_1,t_1,\mathbf{r}_2,t_2) \longrightarrow
ho(\mathbf{r},t)$$

Time Dependent Density Functional Theory (DFT)

allows dynamics and short range/long range correlations



IOP PUBLISHING

J. Phys.: Condens. Matter 24 (2012) 233202 (11pp)

JOURNAL OF PHYSICS: CONDENSED MATTER

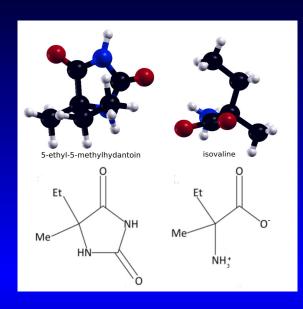
doi:10.1088/0953-8984/24/23/233202

TOPICAL REVIEW

Time-dependent density-functional theory in massively parallel computer architectures: the OCTOPUS project

Xavier Andrade¹, Joseba Alberdi-Rodriguez^{2,3}, David A Strubbe^{4,5}, Micael J T Oliveira⁶, Fernando Nogueira⁶, Alberto Castro⁷, Javier Muguerza³, Agustin Arruabarrena³, Steven G Louie^{4,5}, Alán Aspuru-Guzik¹, Angel Rubio^{2,8} and Miguel A L Marques^{9,10}

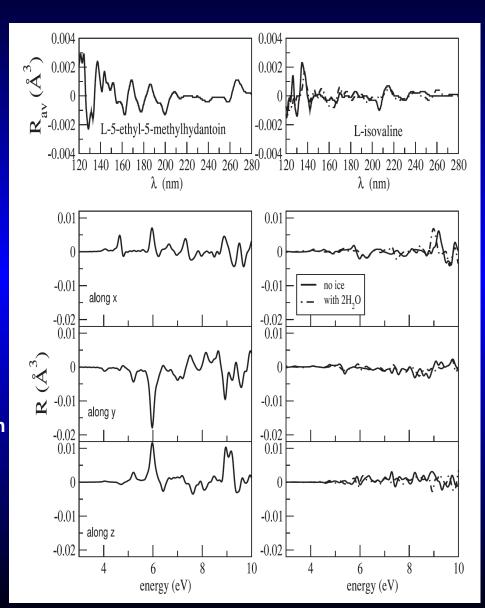
Results

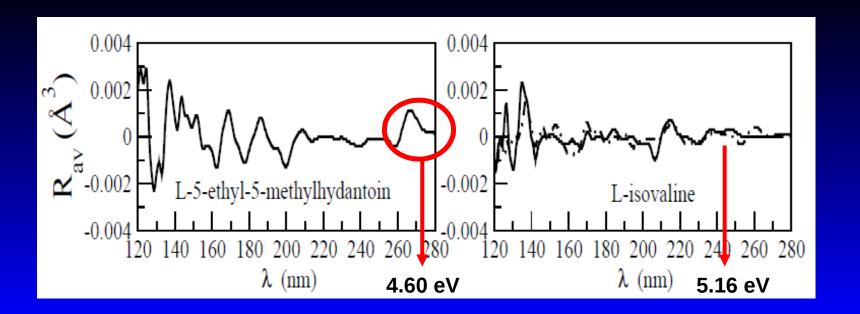


Strong directional dependence on the incidence direction of the light

No big influence of surrounding ice, agreement with film of isovaline

Takahashi J.-I. et al., 2009, Int. J. Mol. Sci., 10, 3044





Peaks for the precursor are overall stronger in a region for selective C-C breaking

Important role of the precursors

In agreement with suggestions from Pizzarello et al, 2012, PNAS USA, 109, 11949

The Vacuum UV region is characterized stronger features

The Vacuum UV CPL light would constitute a P-breaking perturbation inducing a certain asymmetry in the two mirror images for both systems

Information-theoretical content of the precursor and the amino acid

Fisher-Shannon Complexity

$$C_r(FS) = I_r \cdot J_r$$

$$I_r = \int \rho(\mathbf{r}) |\overrightarrow{\nabla} \ln \rho(\mathbf{r})|^2 d^3\mathbf{r}$$

$$J_r = \frac{1}{2\pi e} \exp(\frac{2}{3}S_r)$$

$$J_r = \frac{1}{2\pi e} \exp(\frac{2}{3}S_r) \qquad S_r = -\int \rho(\mathbf{r}) \ln \rho(\mathbf{r}) d^3 \mathbf{r}$$

Departure of the probability density from disorder

Power Entropy Shannon Entropy: measurement of the randomness of the probability density (measure the departure from localizability)

	#ELECTRONS	HARDNESS	C(FS)
amino acid	52	0.18	230
precursor	64	0.25	440

Future Plan

Effect of turbolence

Evaluating the time evolution of complexity (during the chemical reaction while interacting with water)

Thank you!



http://etsf.polytechnique.fr http://www.etsf.eu